



**MICRONOX®R01 and MICRONOX®R02 as exceptionally effective barrier pigments in the design of anticorrosive coatings for highly corrosive environments**



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PROMINDSA RnD Manager



## **INDEX:**

- 1) Company presentation
- 2) Corrosion and prevention
- 3) Micronox R01 and R02: natural pigments with excellent anticorrosion properties

## What we are

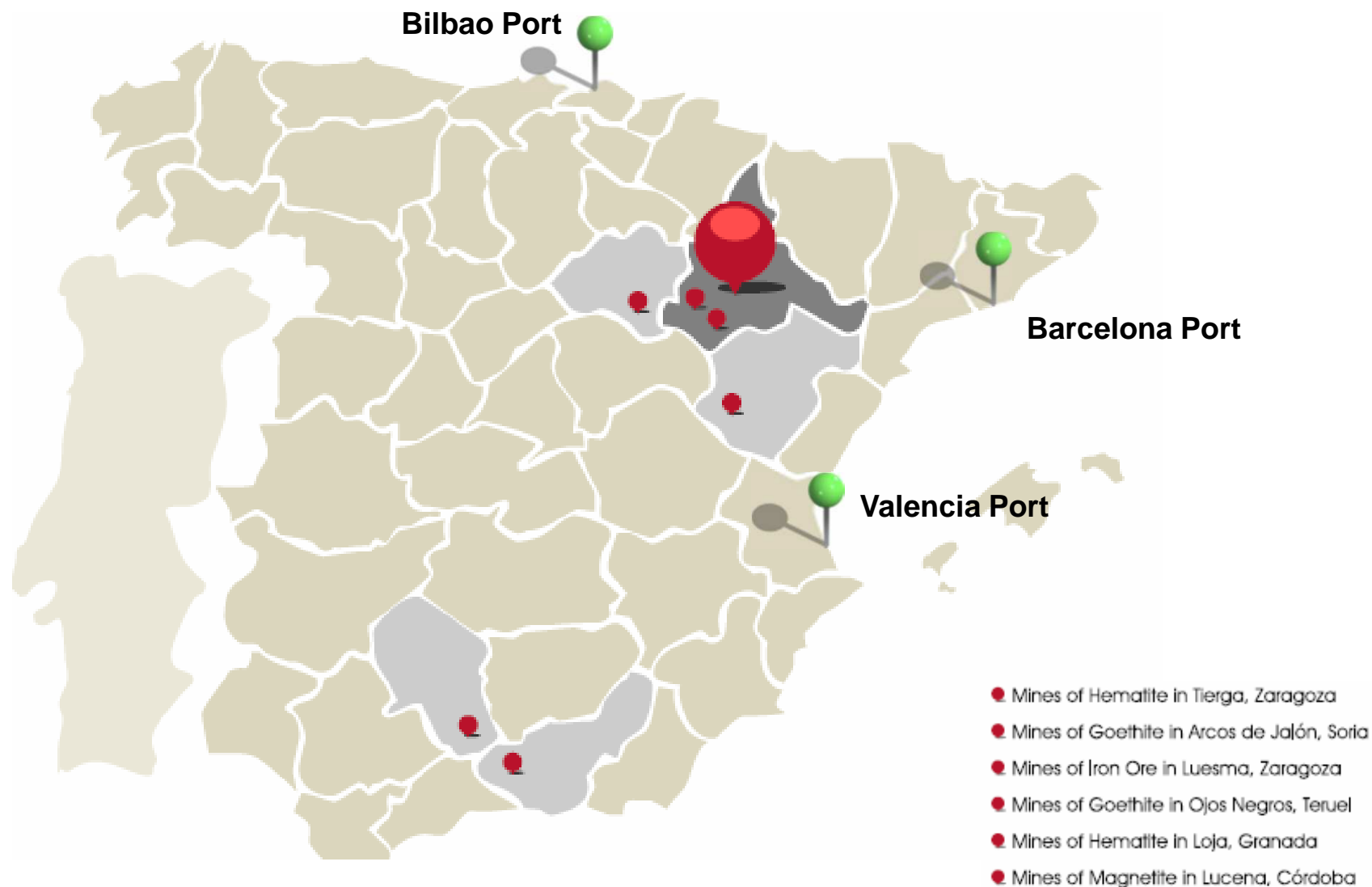
We produce, transform, develop and supply inorganic pigments worldwide.



## Where we are



# Our Factory and Mines



MICRONOX®  
PIGMENTS

- Red Iron Oxide: **Micronox® R**
- Black Iron Oxide: **Micronox® BK**
- Yellow Iron Oxide: **Micronox® Y**
- Brown Iron Oxide: **Micronox® BR**
- Orange Iron Oxide: **Micronox® Marigold**
- **We create and develop customized high quality products**





PROMINDSA



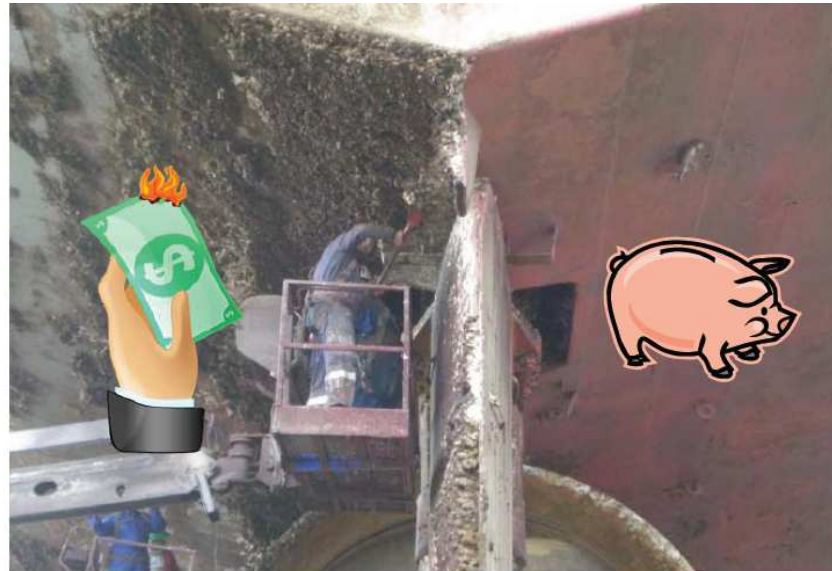
MICRONOX<sup>®</sup>  
PIGMENTS

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The prevention of corrosion is of paramount importance

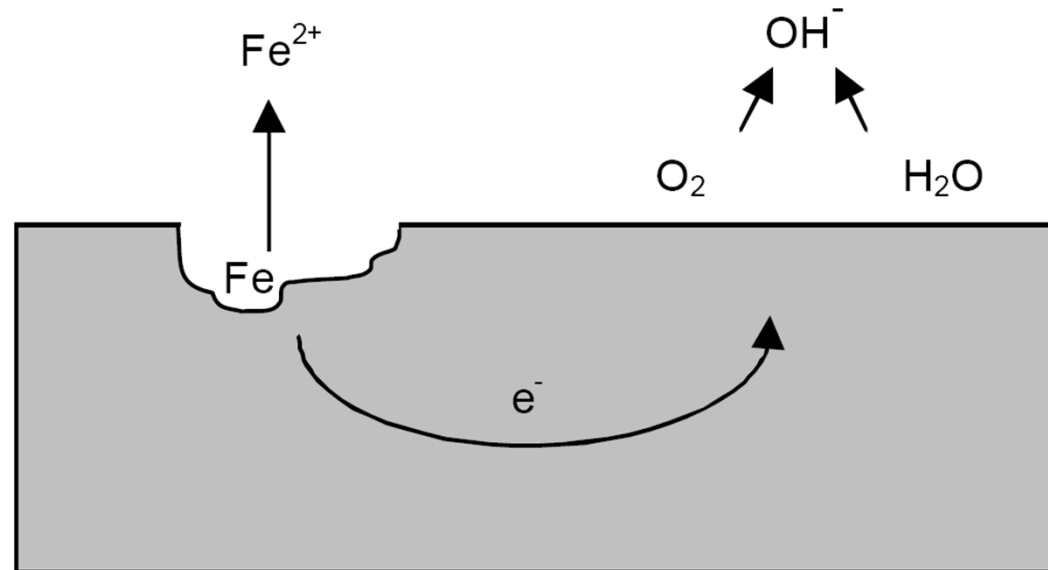
**Corrosion** causes a lot of economic damage.



**Corrosion** can lead to structural failures

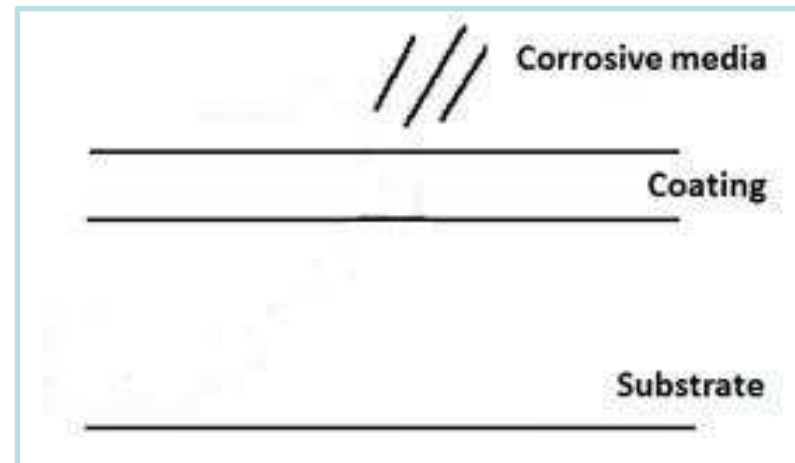


**Corrosion** is an electrochemical process that destroys the surface of metals

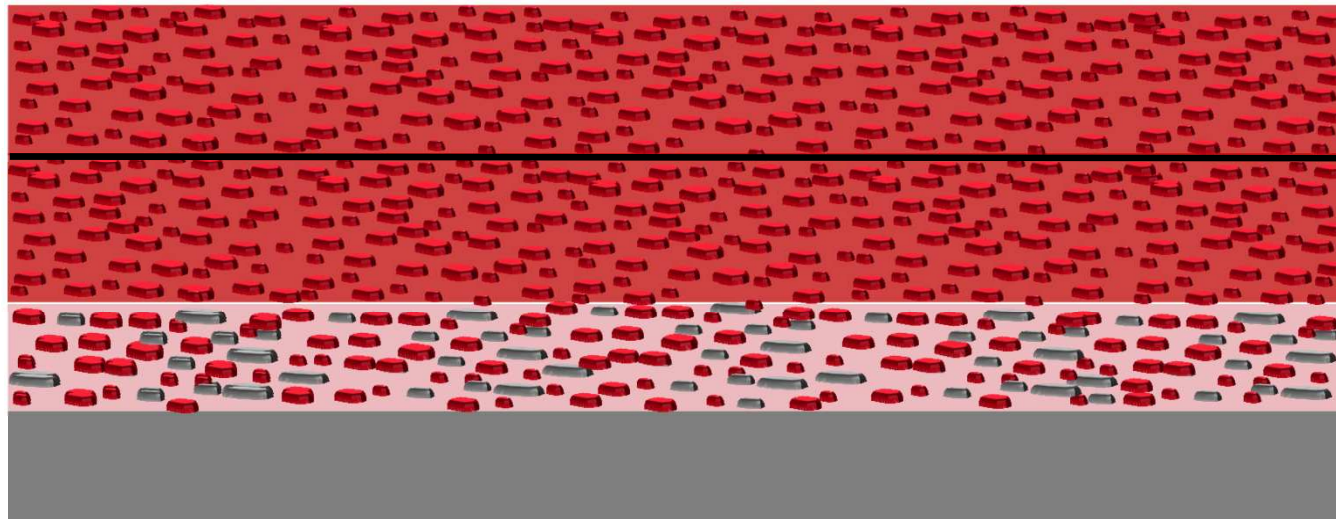


A coating must provide sufficient protection to retard an onset and propagation of corrosion reactions:

- Improve **electrochemical stability** of metal surface (Inhibitive and galvanic effect)
- **Restrict access of corrosive media** to interface ([barrier effect](#))



# Anticorrosion coatings



**Topcoat**  
(color and gloss)

**Buildcoat**  
(Build up thickness)

**Primer**  
(anticorrosion and  
adhesion)

**Substrate**

**Schematic of a typical anticorrosive coating system**



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NATURAL MICRONOX<sup>®</sup> PIGMENTS MANUFACTURING



NOT CHEMICAL PROCESS INVOLVED



Mineral extraction



Non-micronized  $\alpha\text{-Fe}_2\text{O}_3$



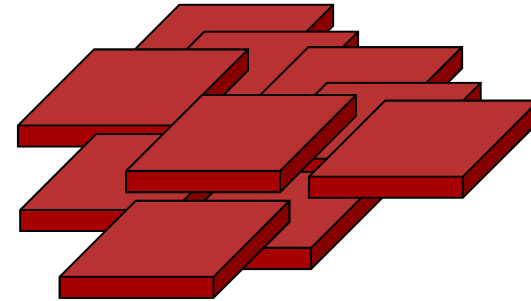
Micronox<sup>®</sup> natural red pigment



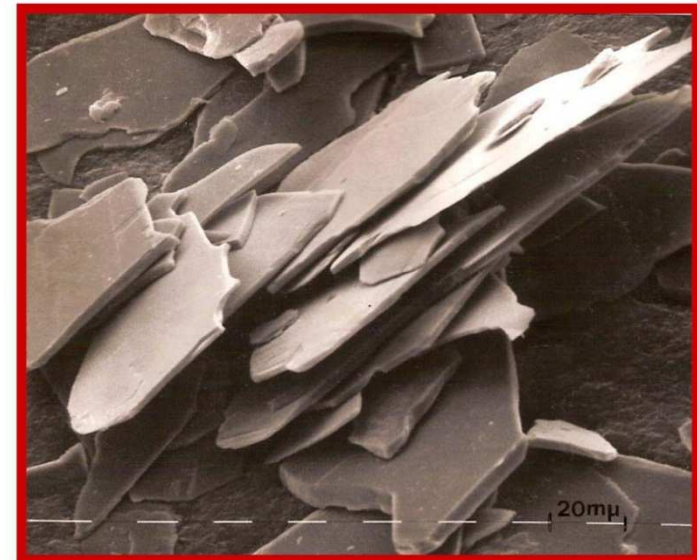
Intensive micronisation & clasifying

## Lamellar Particle shape

- The **natural**  $\alpha\text{-Fe}_2\text{O}_3$  (hematites) grows very slowly (millions of years), thus allowing its crystals to develop its pattern (typical morphology), which is **lamellar**



## MICRONOX<sup>®</sup> PARTICLE SHAPE



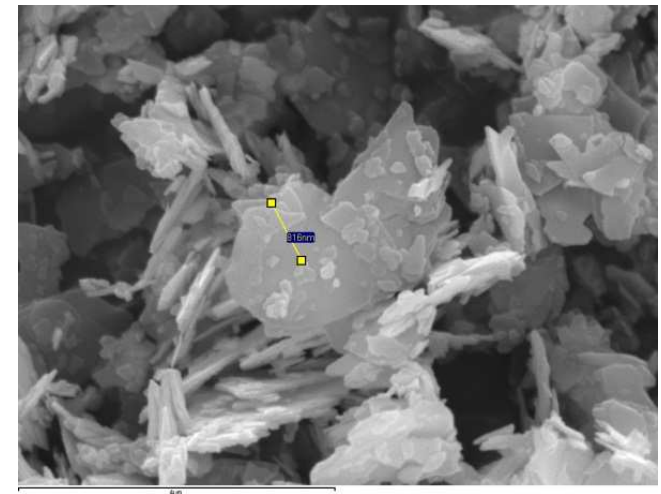
## Lamellar shape



**Iron ore**



**Iron ore after milling**



**Iron ore after micronization**

## SYNTHETIC PIGMENTS MANUFACTURING

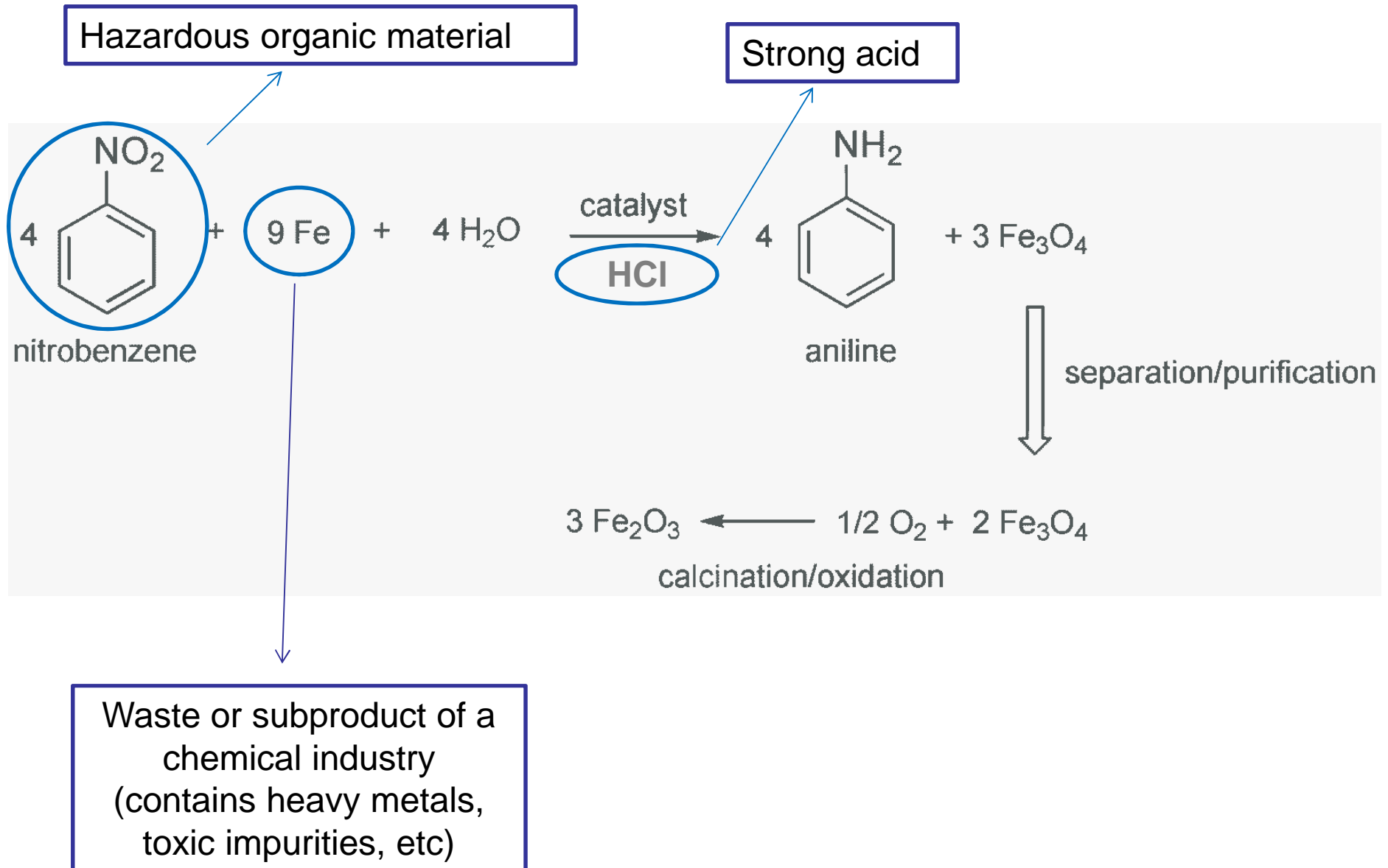


By-  
products/Scrap  
to pigment



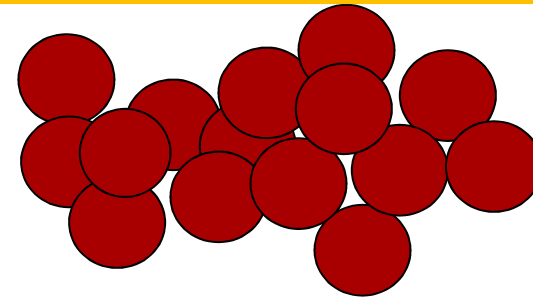
Require the use of **hazardous** and **toxic** substances  
(Strong acids, heavy metals)

# PRODUCTION OF SYNTHETIC IRON OXIDE

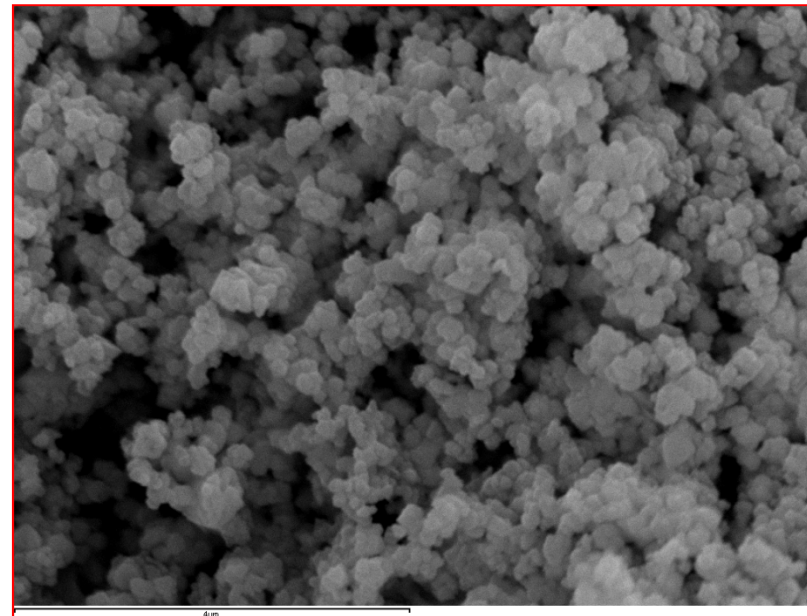


# Equidimensional Particle shape

- The **synthetic**  $\alpha\text{-Fe}_2\text{O}_3$  grows very quickly (a few hours), what prevents the crystals to develop its pattern, and gives rise to more **equidimensional** shaped particles.



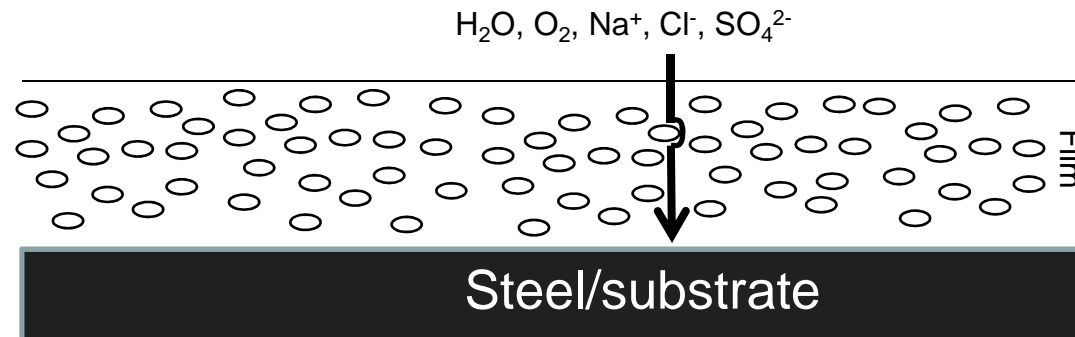
## SYNTHETIC PARTICLE SHAPE



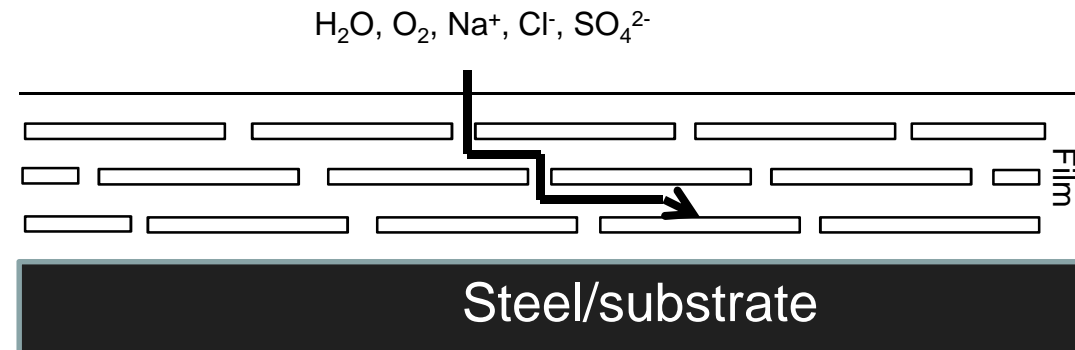
# Barrier effect

Primer coating formulated with **MICRONOX®R01** and **MICRONOX®R02**  
excellent **barrier-effect**

synthetic



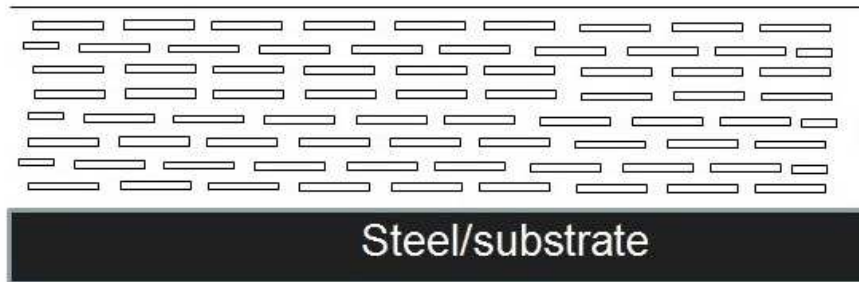
MICRONOX  
R01 and R02





MICRONOX

Micronox lamellas aspect ratio 1:12

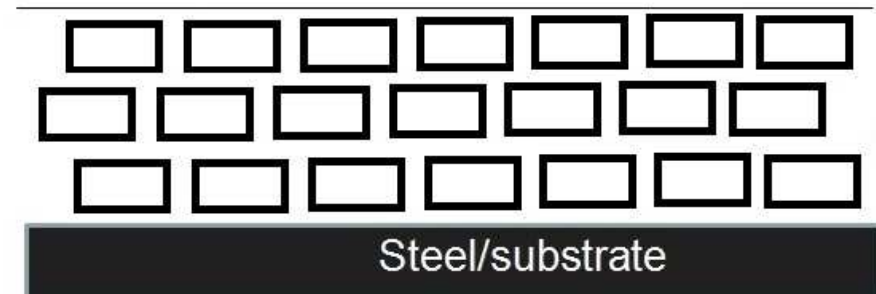


Higher barrier effect



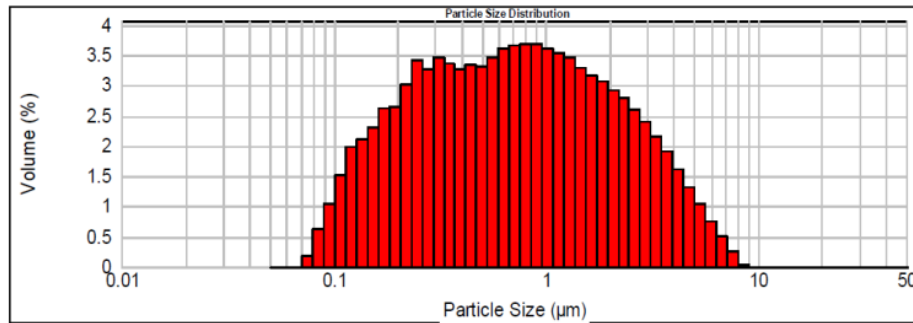
MIO  
(micaceous iron oxide)

MIO lamellas aspect ratio 1:2

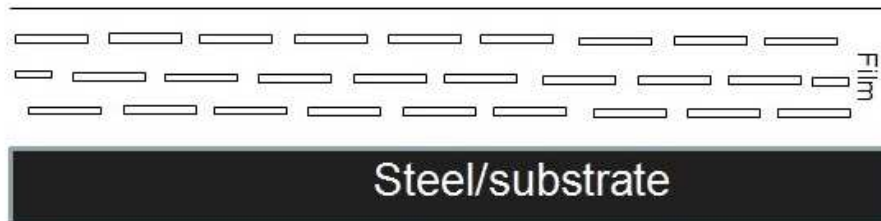


## MICRONOX®R01

High quality of micronization



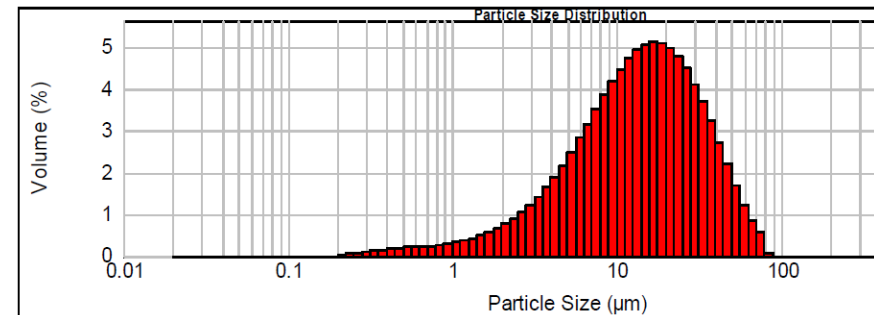
Average particle size: 1.25 µm  
Homogeneous distribution of particle sizes



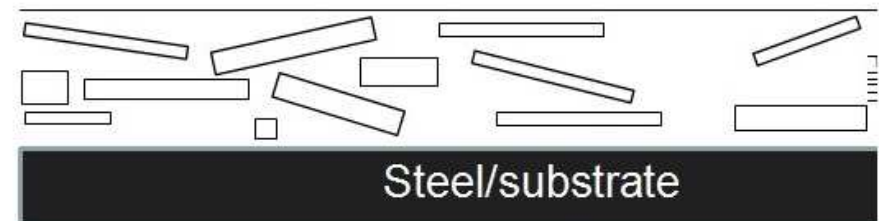
Higher barrier effect

## MICACEOUS IRON OXIDE (MIOX)

Low-medium quality of micronization



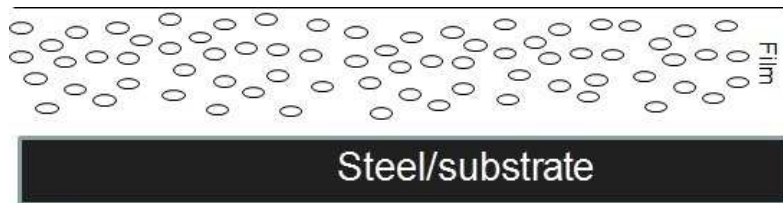
Heterogeneous particle size distribution



Higher permeability

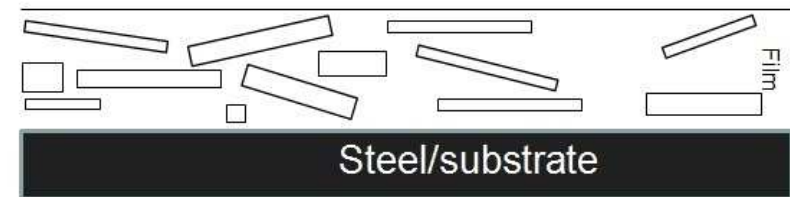
## INFLUENCE OF PARTICLE SIZE (NOT ONLY SHAPE)

synthetic

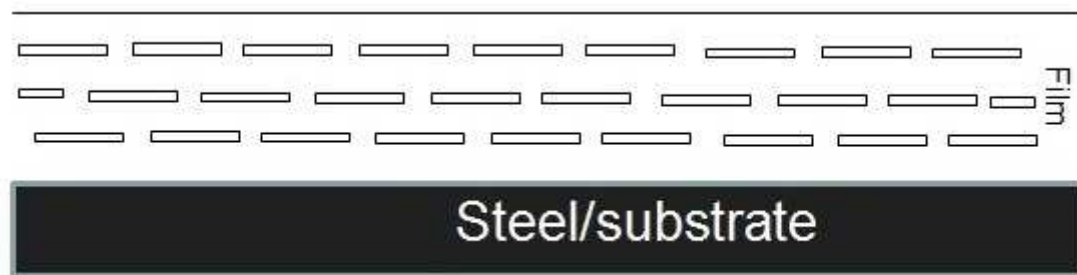


Small particle size-dispersibility

MIOX (natural but low quality  
of micronization)



Lamellar shape-barrier effect

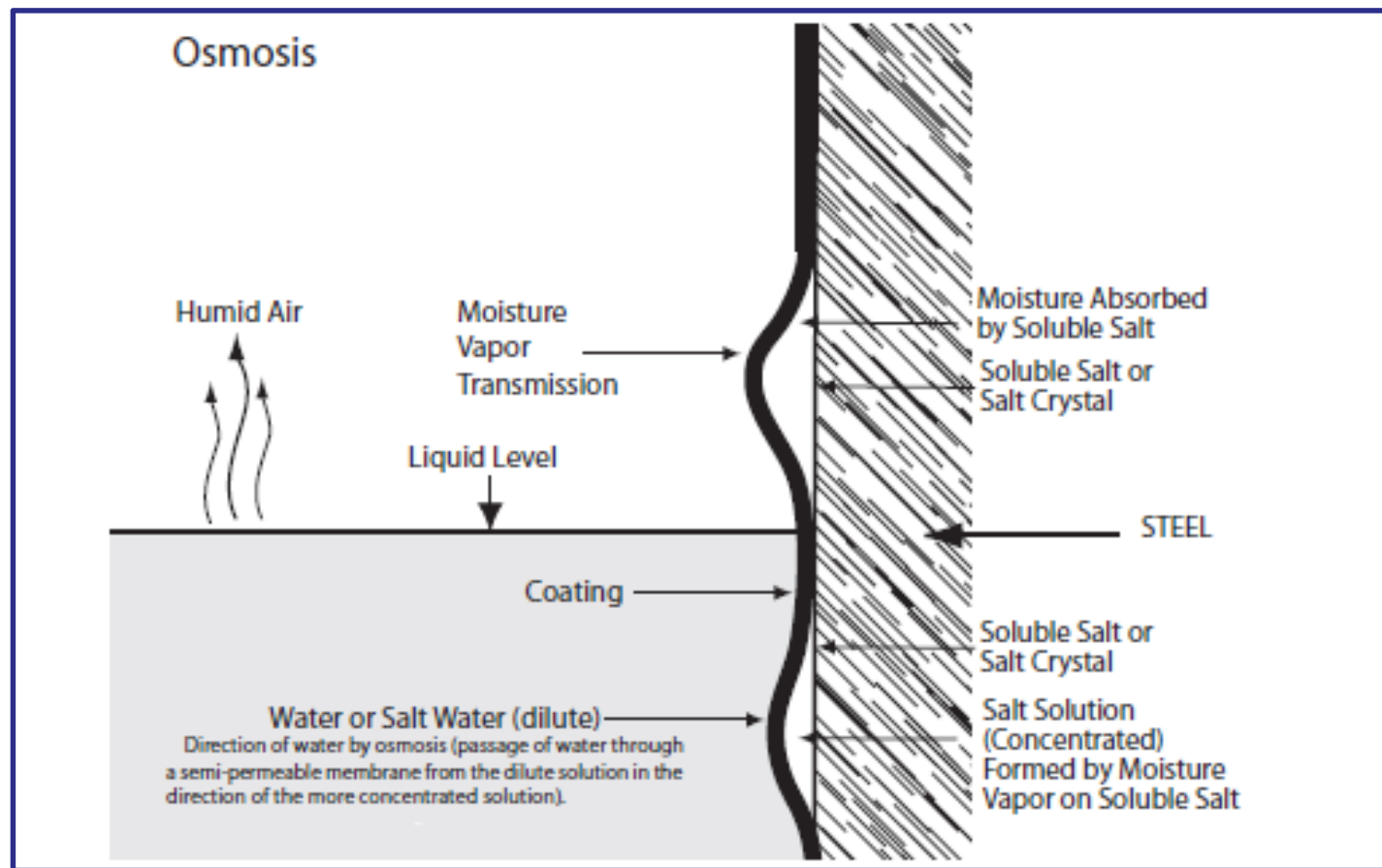


MICRONOX: both small particle size and lamellar shape

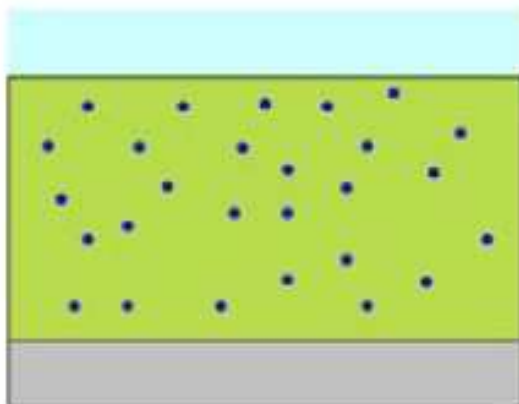
EXCELLENT ANTICORROSION PROPERTIES

**MICRONOX®** Red series: content in **water soluble salts** is **undetectable** ( $< 0.01\%$  wt)  
 Synthetic pigments: typically content of soluble salts is in the range 1.0 - 0.1 % wt

The water uptake produced by **water soluble salts** (typically chloride and sulphate) causes the formation of blisters

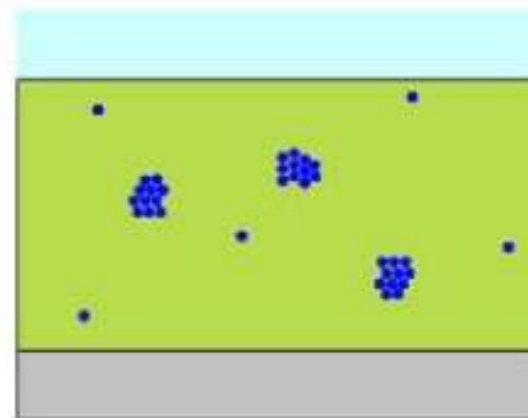


“microfogging”



molecular dispers distributed water

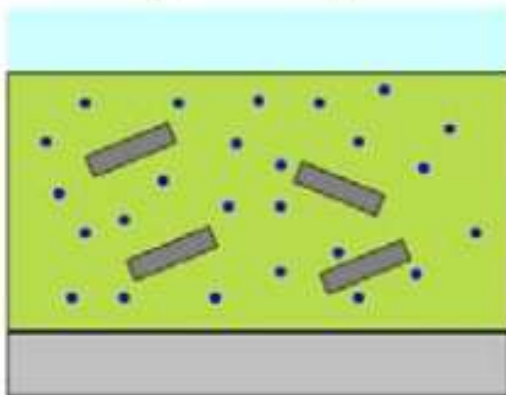
$$T_1 > T_2$$



excess water, formation of micro caverns

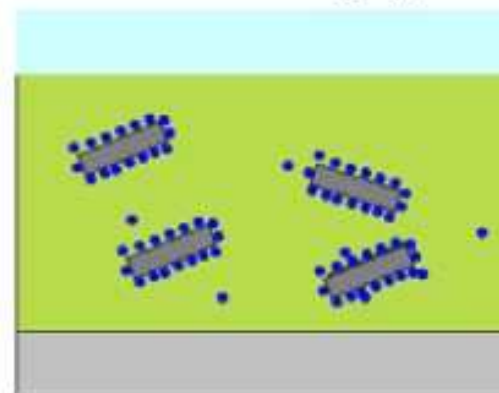
(C.H. Hansen, Prog. Org. Coat. 26 (1995) 113)

durability to temperatur cycling by micaceous iron ore pigments



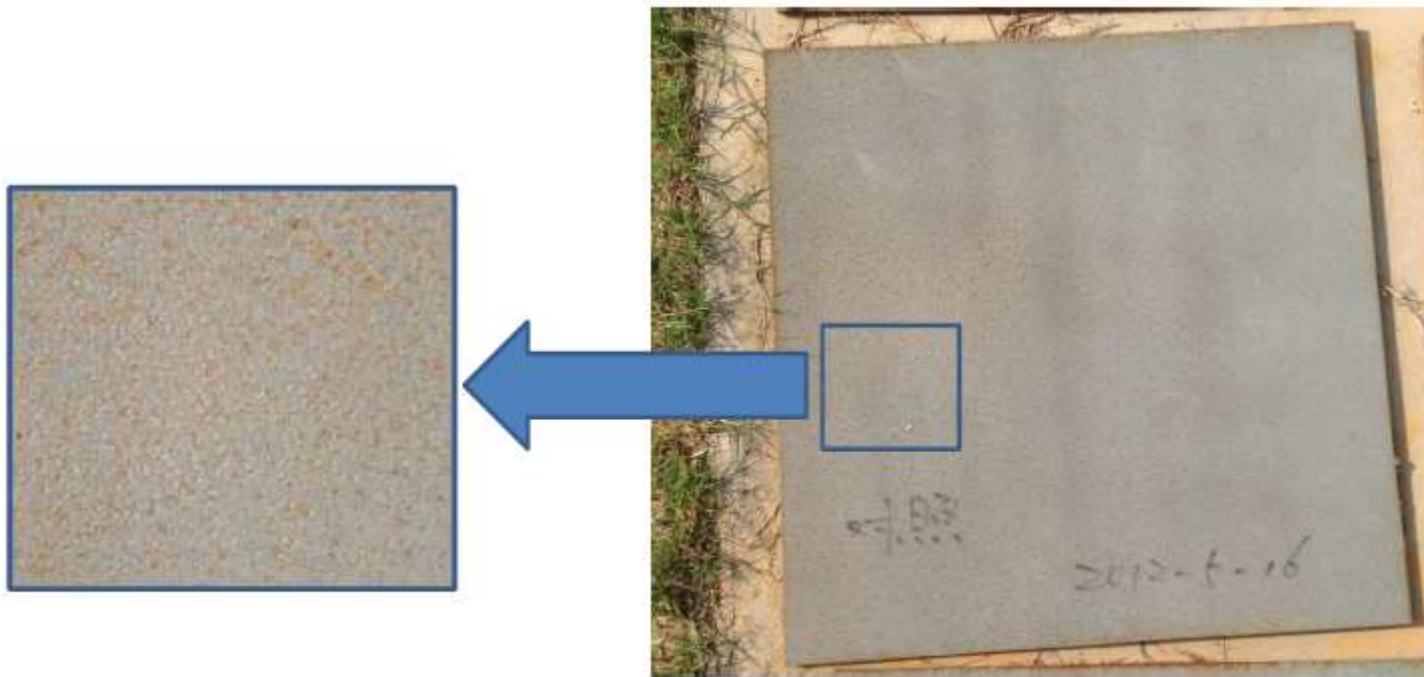
iron ore pigments as condensation nuclei

$$T_1 > T_2$$

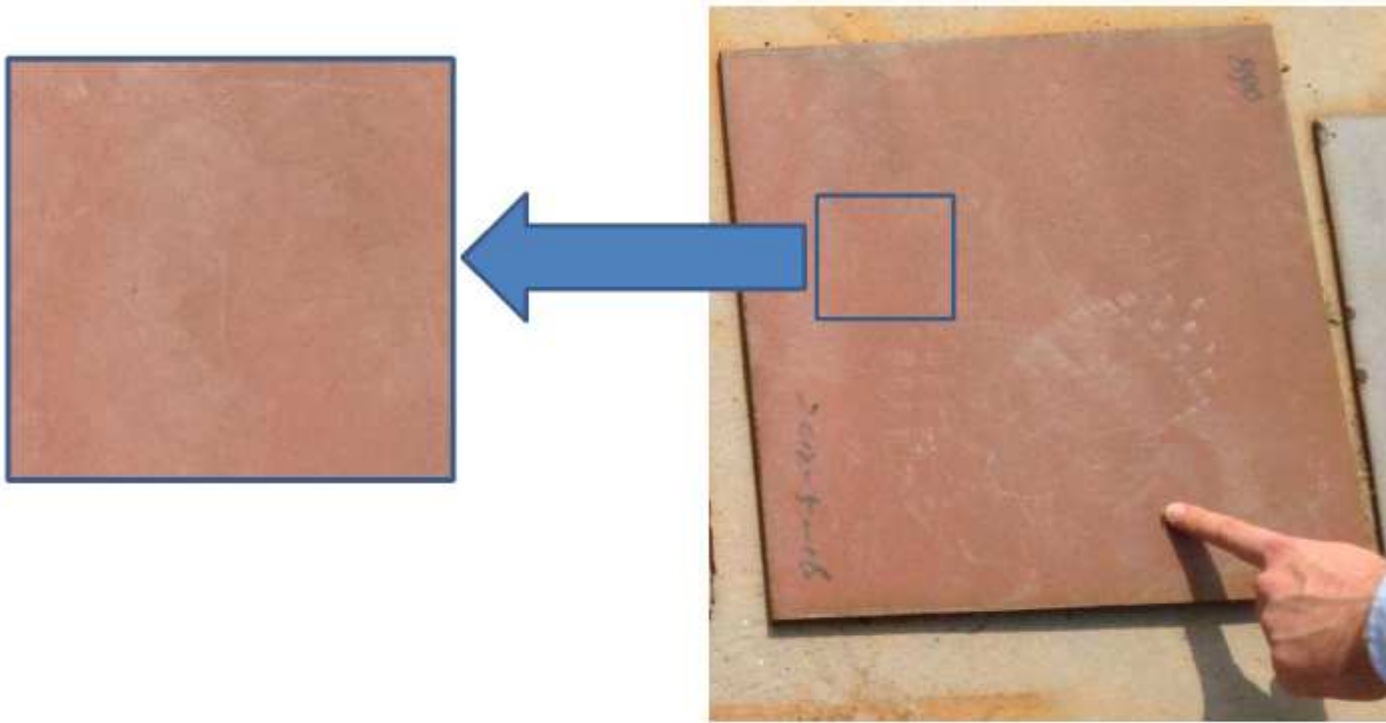


(H. Ochs, J. Vogelsang, Electrochimica Acta, 2004)

## Primer with Zn-based inhibitor pigment



## Primer with Zn-based inhibitor and **MICRONOX<sup>®</sup>R01** pigment

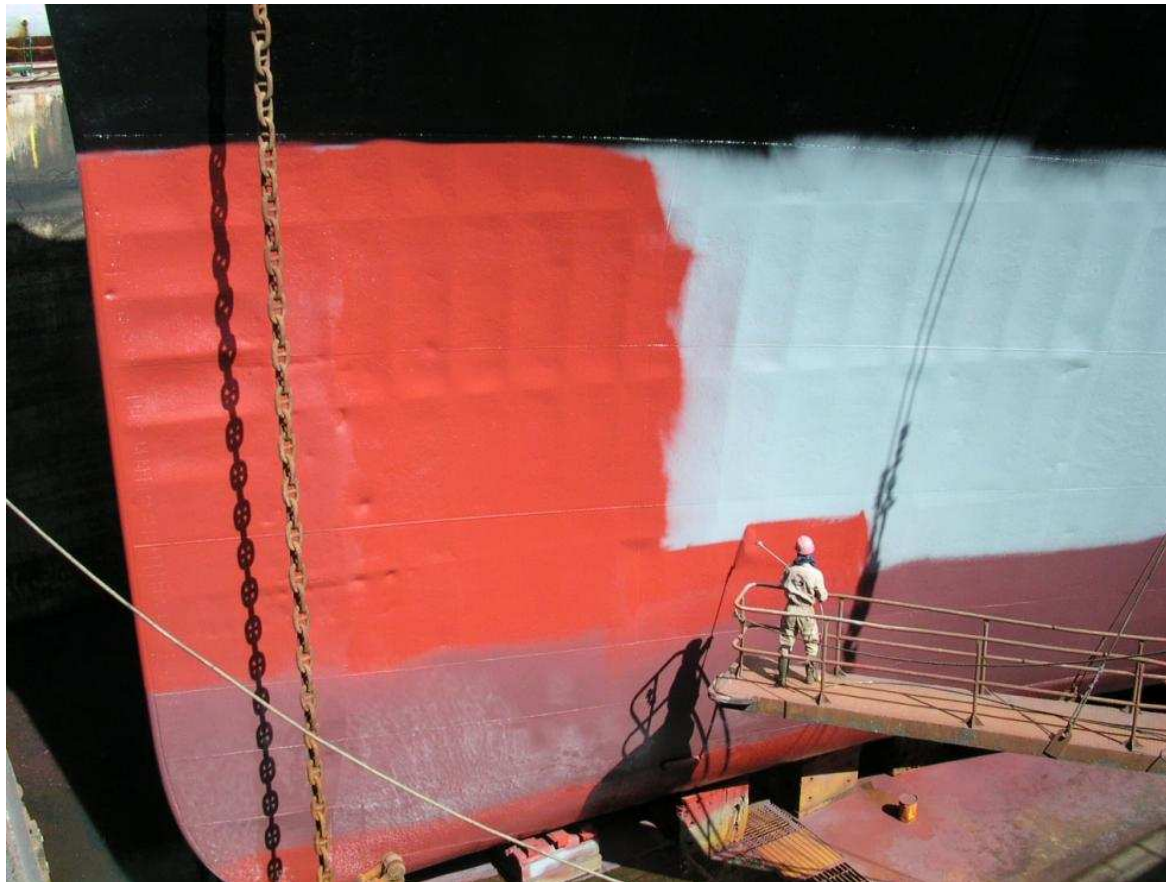


**Synergistic effect**

# Comparison



Micronox<sup>®</sup>R01 and Micronox<sup>®</sup>R02 are  
worldwide standard pigments for  
anticorrosive coatings





Jotun Coatings



**SHERWIN-WILLIAMS.**





спасибо

Thank you

Gracias